

(+)- and (-)-*syn*-2-Isobutyl-4-methylazetidene-2,4-dicarboxylic Acids from the Extract of *Monascus pilosus*-Fermented Rice (Red-Mold Rice)

Toshihiro Akihisa,^{*,†} Shoichi Mafune,[†] Motohiko Ukiya,[†] Yumiko Kimura,[‡] Ken Yasukawa,[‡] Takashi Suzuki,[‡] Harukuni Tokuda,[§] Nobukazu Tanabe,[⊥] and Tadahiko Fukuoka[⊥]

College of Science and Technology, Nihon University, 1-8 Kanda Surugadai, Chiyoda-ku, Tokyo 101-8308, Japan, College of Pharmacy, Nihon University, 7-7-1 Narashinodai, Funabashi-shi, Chiba 274-8555, Japan, Department of Biochemistry, Kyoto Prefectural University of Medicine, Kamigyo-ku, Kyoto 602-0841, Japan, and Research Center, Gunze, Co., Ishiburo 1, Ikurashinmachi, Ayabe-shi, Kyoto 623-8512, Japan

Received August 26, 2003

The structures of two enantiomeric azetidene-type amino acids isolated from the *n*-butanol-soluble fraction of the 70% ethanol extract of red-mold rice fermented with *Monascus pilosus* were established to be (+)-[**1**; (+)-monascumic acid] and (-)-*syn*-2-isobutyl-4-methylazetidene-2,4-dicarboxylic acids [**2**; (-)-monascumic acid] based on spectroscopic methods.

Species of the fungi *Monascus* (Eurotiaceae) have been utilized for making fermented food and preserving meat for hundreds of years. Red-mold rice fermented using *Monascus* spp. is effective in decreasing blood pressure¹ and lowering plasma cholesterol levels^{2,3} and has antibacterial activity.⁴ γ -Aminobutyric acid (GABA), which possesses anti-hypertensive effects in humans, has been isolated from red-mold rice.⁵ Endo³ discovered that *Monascus ruber* produces monacolin K (lovastatin; mevinolin), an active methylated form of compactin, in liquid fermentation. Monacolin K functions as an inhibitor of 3-hydroxy-3-methylglutaryl-coenzyme A (HMG-CoA) reductase, which is the regulatory and rate-limiting enzyme of cholesterol biosynthesis.⁶ The fact that red-mold rice can suppress the synthesis of cholesterol has also been confirmed.⁷ In the course of our search for potential antitumor promoters from natural sources,⁸ we were especially interested to undertake the investigation of red-mold rice constituents.⁹ In this paper, we report the isolation and characterization of two enantiomeric azetidene (trimethyleneimine)-type amino acids, **1** and **2**, from the 70% ethanol extract of red-mold rice fermented with *Monascus pilosus*.

The molecular formula of compound **1** was determined as C₁₀H₁₇NO₄ from the HREIMS ([M]⁺ *m/z* 215.1157) as well as from its ¹³C NMR DEPT. The compound has two secondary methyls [δ_{H} 0.81 (d, *J* = 6.4 Hz), 0.88 (d, *J* = 6.6 Hz); solvent: DMSO-*d*₆], one tertiary methyl [δ_{H} 1.31 (s)], two methylenes [δ_{C} 43.7 (t), δ_{H} 1.36 (1H, dd, *J* = 2.9, 12.8 Hz) and 1.71 (1H, dd, *J* = 9.0, 12.8 Hz); and δ_{C} 45.0 (t), δ_{H} 1.99 (1H, d, *J* = 16.3 Hz) and 2.45 (1H, d, *J* = 16.3 Hz)], one methine [δ_{C} 23.7 (d), δ_{H} 1.66 (m)], two sp³ quaternary carbons [δ_{C} 72.2 (t) and 77.6 (t)] adjacent to a secondary amine group [δ_{H} 1.91 (1H, s); ν_{max} 3422, 3290 cm⁻¹], and two carboxyls [ν_{max} 1717, 1681, 1239 cm⁻¹; δ_{C} 172.8 (s) and 173.6 (s); δ_{H} 7.94 (2H, s)]. These data, in combination with diagnostic MS fragment ions at *m/z* 170 [M - COOH]⁺, 158 [M - C₄H₉ (isobutyl)]⁺, and 154 [M - COOH - CH₃ - H]⁺, suggested that **1** possesses a four-membered azetidene (trimethyleneimine) ring substituted with carboxyl and isobutyl groups at C-2 and carboxyl and

methyl groups at C-4. The proposed structure of **1** was supported by the analysis of ¹³C DEPT NMR, ¹H-¹H COSY, HMQC, and HMBC spectra. The relative configuration of **1** was established by NOESY and difference NOE experiments. Compound **1** showed significant NOE correlations between H_a-6 (δ_{H} 1.36) and H_b-3 (δ_{H} 2.45)-H-5 (δ_{H} 1.31) (Figure 1), which suggested that the isobutyl group at C-2 and the methyl group at C-4 were oriented on the same face of the azetidene ring. We concluded that **1** is *syn*-2-isobutyl-4-methylazetidene-2,4-dicarboxylic acid, and since **1** exhibited positive specific optical rotation ($[\alpha]_{\text{D}}^{25} +3.7^{\circ}$), we named it (+)-monascumic acid.

Compound **2**, which has the same molecular formula C₁₀H₁₇NO₄ (HREIMS [M]⁺ *m/z* 215.1160; HRFABMS [M + H]⁺ *m/z* 216.1235) as **1**, exhibited EIMS, IR, and ¹³C and ¹H NMR spectral data, melting point data (Experimental Section), and NOE correlations (Figure 1) indistinguishable from those of **1**, suggesting that **2** was an enantiomer of **1** and possessed the structure *syn*-2-isobutyl-4-methylazetidene-2,4-dicarboxylic acid. We named **2** as (-)-monascumic acid since it exhibited an almost opposite specific rotation ($[\alpha]_{\text{D}}^{25} -4.4^{\circ}$) of that for compound **1**.

The ¹³C and ¹H NMR spectral data and the NOE correlations of compounds **1** and **2** determined in pyridine-*d*₅ were fully consistent with their proposed structures.

The two enantiomeric azetidene-type amino acids, (+)-(**1**) and (-)-monascumic acids (**2**), isolated from the 70% EtOH extract of red-mold rice fermented with *M. pilosus* in this study are the new naturally occurring compounds. The occurrence of the azetidene ring system in natural products is uncommon, and derivatives of this four-membered moiety have been isolated so far only from the roots and leaves of *Convallaria majalis* (lily-of-the-valley),¹⁰ the culture broth of *Streptomyces cacaoi*,¹¹ the roots of barley,¹² and the Okinawan marine sponge *Penares* sp.¹³ The former three azetidines^{10–12} and monascumic acids (**1** and **2**), isolated in this study, are the α -amino acids or their derivatives possessing a secondary amine group.

Experimental Section

General Experimental Procedures. Crystallizations were performed in ethyl acetate (EtOAc), and melting points were determined on a Yanagimoto micro melting point apparatus and are uncorrected. Optical rotations were measured on a JASCO P-1030 polarimeter in acetone at 25 °C. IR spectra were recorded on a JASCO IR-300 spectrometer in KBr disks.

* To whom correspondence should be addressed. Fax: +81-3-3293-7572. E-mail: akihisa@chem.cst.nihon-u.ac.jp.

[†] College of Science and Technology, Nihon University.

[‡] College of Pharmacy, Nihon University.

[§] Kyoto Prefectural University of Medicine.

[⊥] Gunze, Co.

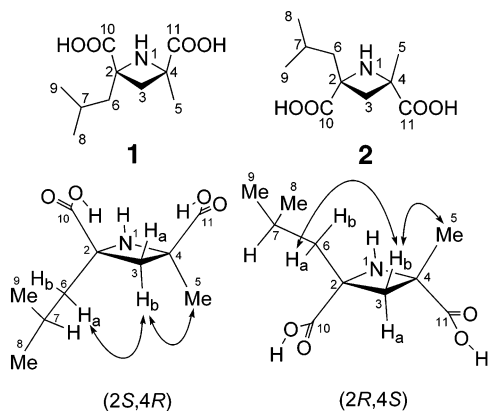


Figure 1. Structures and major NOE correlations (\longleftrightarrow) for **1** [(2*S*,4*R*) or (2*R*,4*S*)] and **2** [(2*R*,4*S*) or (2*S*,4*R*)].

NMR spectra were recorded with a JEOL LA-400 spectrometer at 400 MHz (^1H NMR) and 100 MHz (^{13}C NMR) in $\text{DMSO-}d_6$ or in pyridine- d_5 with tetramethylsilane as internal standard. Electron-impact mass spectra (EIMS; 70 eV) and high-resolution EIMS (HREIMS) were recorded on a JEOL JMS-BU20 spectrometer using a direct inlet system. HRFABMS were obtained with a JEOL JMS-BU20 spectrometer using glycerol as the matrix. Octadecyl silica (Chromatorex-ODS, 100–200 mesh; Fuji Silysia Chemical, Ltd., Aichi, Japan) was used for open column chromatography. Reversed-phase preparative HPLC was carried out on a 25 cm \times 10 mm i.d. Pegasil ODS II (Senshu Scientific Co., Ltd., Tokyo, Japan) C_{18} silica column, at 25 $^\circ\text{C}$ with $\text{H}_2\text{O}-\text{CH}_3\text{CN}-\text{acetic acid}$ (90:10:3, v/v/v) as mobile phase at 3 mL/min. A refractive index detector was used for reversed-phase HPLC.

Materials. Red-mold rice was prepared as follows.⁵ Well-milled rice was immersed in H_2O for 12 h, removed from the H_2O , and autoclaved for 30 min at 121 $^\circ\text{C}$. *M. pilosus* IFO 4520 obtained from the Institute for Fermentation (IFO), Osaka, Japan, was inoculated into the cooked paddy rice and cultured for 14 days at 30 $^\circ\text{C}$ under aerobic conditions. The rice was dried at 70 $^\circ\text{C}$ to about 8% moisture content.

Extraction and Fractionation. Red-mold rice (1.5 kg) was extracted with 12 L of 70% EtOH for 30 min under stirring. The mixture was suction-filtered, and the residue was washed with 3 L of 70% EtOH. The combined filtrate and wash were evaporated in vacuo at 50 $^\circ\text{C}$ to leave the 70% EtOH extract (152.9 g). The extract was suspended in 1 L of H_2O , and the suspension was extracted with EtOAc (5 times with 0.5 L each) and then with *n*-butanol (*n*-BuOH) (5 times with 0.5 L each). The EtOAc and *n*-BuOH solutions and the remaining H_2O phase were evaporated in vacuo at 50 $^\circ\text{C}$ to yield the EtOAc (26.1 g), *n*-BuOH (23.2 g), and H_2O (100.4 g) soluble fractions.

Isolation. A portion of the *n*-BuOH fraction (20.0 g) was subjected to chromatography on an octadecyl silica column (100 g). Stepwise gradient elution of the column with a mixture of solvents yielded the following 10 fractions with a descending order of polarity: fractions A [9.67 g; MeOH (M)– H_2O (H) (3:7, v/v) 1.5 L], B [1.76 g; M–H (3:7) 2.0 L], C [0.35 g; M–H (3:7) 4.0 L and M–H (7:3) 1.0 L], D [2.64 g; M–H (7:3) 3.0 L], E [0.56 g; M–H (7:3) 6.0 L and M–H (9:1) 1.0 L], F [0.46 g; M–H (9:1) 1.0 L], G [0.58 g; M–H (9:1) 2.0 L and M 0.5 L], H [0.59 g; M 2.5 L], I [0.44 g; M 2.5 L and M–EtOAc (E) (9:1) 3.5 L], and J [0.44 g; M–E (1:1) 1.5 L and E 2.0 L]. Preparative HPLC of a portion of fraction B (45 mg) yielded compounds **1** [10 mg; retention time (t_R) 30 min] and **2** (30 mg; t_R 26 min).

(+)-Monascumic Acid (1): colorless needles, mp 122–123 $^\circ\text{C}$; $[\alpha]_D^{25} +3.7^\circ$ (*c* 0.39, acetone); IR ν_{max} 3420 and 3290 ($>\text{NH}$), 1717, 1681, and 1239 ($-\text{COOH}$) cm^{-1} ; ^1H NMR ($\text{DMSO-}d_6$, 400 MHz) δ 7.95 (2H, s, 2 \times COOH, s), 2.45 (1H, d, $J = 16.3$ Hz, H_b -3), 1.98 (1H, d, $J = 16.3$ Hz, H_a -3), 1.91 (1H, s, H-1), 1.70 (1H, dd, $J = 8.8$, 12.2 Hz, H_b -6), 1.66 (1H, m, H-7), 1.36 (1H, dd, $J = 2.2$, 12.2 Hz, H_a -6), 1.31 (3H, s, H-5), 0.87 (3H, d, $J = 6.4$ Hz, H-8 or H-9), 0.81 (3H, d, $J = 6.1$ Hz, H-9 or H-8); ^{13}C NMR ($\text{DMSO-}d_6$, 100 MHz) δ 173.4 (C, C-11), 172.8 (C, C-10),

77.6 (C, C-4), 72.2 (C, C-2), 45.0 (CH_2 , C-3), 43.7 (CH_2 , C-6), 24.8 (CH_3 , C-8 or C-9), 23.7 (CH, C-7), 21.9 (CH_3 , C-9 or C-8), 21.7 (CH_3 , C-5); ^1H NMR (pyridine- d_5 , 400 MHz) δ 9.49 (2H, 2 \times COOH, s), 2.94 (1H, d, $J = 16.4$ Hz, H_b -3), 2.76 (1H, d, $J = 16.4$ Hz, H_a -3), 2.34 (1H, dd, $J = 8.3$, 12.4 Hz, H_b -6), 2.11 (1H, m, H-7), 1.84 (3H, s, H-5), 1.76 (1H, br d, $J = 12.4$ Hz, H_a -6), 1.15 (3H, d, $J = 6.6$ Hz, H-8 or H-9), 1.01 (3H, d, $J = 6.6$ Hz, H-9 or H-8); ^{13}C NMR (pyridine- d_5 , 100 MHz) δ 174.8 (2C, CH_3 , C-10 and C-11), 79.0 (C, C-4), 74.5 (C, C-2), 46.4 (CH_2 , C-3), 45.1 (CH_2 , C-6), 25.3 (CH, C-7), 25.1 (CH_3 , C-8 or C-9), 22.5 (CH_3 , C-9 or C-8), 22.3 (CH_3 , C-5); NOESY and difference NOE experiments showed significant NOE correlations between H_a -6 (δ 1.36 in $\text{DMSO-}d_6$, 1.76 in pyridine- d_5) and H_b -3 (δ 2.45, 2.94)–H-5 (δ 1.31, 1.84); EIMS m/z 215 [$\text{M}]^+$ (6), 197 [$\text{M} - \text{H}_2\text{O}]^+$ (26), 172 [$\text{M} - \text{C}_3\text{H}_7$ (isopropyl)] $^+$ (41), 170 [$\text{M} - \text{COOH}]^+$ (16), 158 [$\text{M} - \text{C}_4\text{H}_9$ (isobutyl)] $^+$ (6), 154 [$\text{M} - \text{COOH} - \text{CH}_3 - \text{H}]^+$ (81), 141 (6), 130 (35), 112 [$\text{M} - \text{C}_4\text{H}_9 - \text{COOH} - \text{H}]^+$ (23), 88 (100), 85 (35), 70 (17), 58 (35); HREIMS m/z 215.1157 (calcd for $\text{C}_{10}\text{H}_{17}\text{NO}_4$, 215.1157).

(-)-Monascumic Acid (2): colorless needles, mp 122–123 $^\circ\text{C}$; $[\alpha]_D^{25} -4.4^\circ$ (*c* 0.36, acetone); IR ν_{max} 3422 and 3290 ($>\text{NH}$), 1718, 1682, and 1239 ($-\text{COOH}$) cm^{-1} ; ^1H NMR ($\text{DMSO-}d_6$, 400 MHz) δ 7.95 (2H, s, 2 \times COOH, s), 2.45 (1H, d, $J = 16.3$ Hz, H_b -3), 1.99 (1H, d, $J = 16.3$ Hz, H_a -3), 1.91 (1H, s, H-1), 1.71 (1H, dd, $J = 9.0$, 12.8 Hz, H_b -6), 1.66 (1H, m, H-7), 1.36 (1H, dd, $J = 2.9$, 12.8 Hz, H_a -6), 1.31 (3H, s, H-5), 0.88 (3H, d, $J = 6.6$ Hz, H-8 or H-9), 0.81 (3H, d, $J = 6.4$ Hz, H-9 or H-8); ^{13}C NMR ($\text{DMSO-}d_6$, 100 MHz) δ 173.6 (C, C-11), 172.8 (C, C-10), 77.6 (C, C-4), 72.2 (C, C-2), 45.0 (CH_2 , C-3), 43.7 (CH_2 , C-6), 24.8 (CH_3 , C-8 or C-9), 23.7 (CH, C-7), 21.9 (CH_3 , C-9 or C-8), 21.6 (CH_3 , C-5); ^1H NMR (pyridine- d_5 , 400 MHz) δ 9.40 (2H, 2 \times COOH, s), 2.93 (1H, d, $J = 16.0$ Hz, H_b -3), 2.76 (1H, d, $J = 16.0$ Hz, H_a -3), 2.33 (1H, dd, $J = 8.0$, 12.8 Hz, H_b -6), 2.10 (1H, m, H-7), 1.84 (3H, s, H-5), 1.74 (1H, br d, $J = 12.8$ Hz, H_a -6), 1.13 (3H, d, $J = 6.6$ Hz, H-8 or H-9), 1.00 (3H, d, $J = 6.6$ Hz, H-9 or H-8); ^{13}C NMR (pyridine- d_5 , 100 MHz) δ 174.9 (2C, CH_3 , C-10 and C-11), 79.0 (C, C-4), 74.5 (C, C-2), 46.4 (CH_2 , C-3), 45.1 (CH_2 , C-6), 25.3 (CH, C-7), 25.1 (CH_3 , C-8 or C-9), 22.6 (CH_3 , C-9 or C-8), 22.3 (CH_3 , C-5); NOESY and difference NOE experiments showed significant NOE correlations between H_a -6 (δ 1.36 in $\text{DMSO-}d_6$, 1.74 in pyridine- d_5) and H_b -3 (δ 2.45, 2.93)–H-5 (δ 1.31, 1.84); EIMS m/z 215 [$\text{M}]^+$ (16), 197 (67), 172 (80), 170 (45), 158 (16), 154 (100), 141 (13), 130 (70), 112 (42), 88 (97), 85 (61), 70 (25), 58 (52); HREIMS m/z 215.1160 (calcd for $\text{C}_{10}\text{H}_{17}\text{NO}_4$, 215.1157); HRFABMS m/z 216.1236 (calcd for $\text{C}_{10}\text{H}_{18}\text{NO}_4$, 216.1235).

Acknowledgment. This work was supported, in part, by a grant from the Ministry of Education, Culture, Sports, Science, and Technology, Japan, to promote multidisciplinary research projects, and by a Grant-in-Aid from the Ministry of Education, Culture, Sports, Science, and Technology, Japan.

References and Notes

- Tsuji, K.; Ichikawa, T.; Tanabe, N.; Abe, S.; Tarui, S.; Nakagawa, Y. *Nippon Noeigakagaku Kaishi* **1992**, 66, 1241–1246.
- Endo, A. *J. Antibiot.* **1979**, 32, 852–854.
- Endo, A. *J. Med. Chem.* **1985**, 28, 401–405.
- Wong, H. C.; Bau, Y. S. *Plant Physiol.* **1977**, 60, 578–581.
- Kohama, Y.; Matsumoto, S.; Mimura, T.; Tanabe, N.; Inada, A.; Nakanishi, T. *Chem. Pharm. Bull.* **1987**, 35, 2484–2489.
- Albert, A. W. *Am. J. Cardiol.* **1988**, 62, 10J–15J.
- Martinokova, L.; Juzlova, P.; Vesely, D. *J. Appl. Bacteriol.* **1995**, 79, 609–616.
- Akihisa, T.; Yasukawa, K. In *Studies in Natural Products Chemistry, Vol. 25. Bioactive Natural Products (Part F)*; Atta-ur-Rahman, Ed.; Elsevier Science B.V.: Amsterdam, 2001; pp 43–87.
- Ukiya, M.; Akihisa, T.; Tokuda, H.; Mafune, S.; Tanabe, N.; Fukuoka, T.; Nishino, H. *Abstracts for the Papers presented at the 7th Annual Meeting of the Japanese Society for Food Factors (JSOFF)*; 2002; p 41.
- Cromwell, N. H.; Phillips, B. *Chem. Rev.* **1979**, 79, 331–358.
- Isono, K.; Asahi, K.; Suzuki, S. *J. Am. Chem. Soc.* **1969**, 91, 7490–7505.
- Nomoto, K.; Mino, Y.; Ishida, T.; Yoshioka, H.; Ota, N.; Inoue, M.; Takagai, S.; Takemoto, T. *J. Chem. Soc., Chem. Commun.* **1981**, 338–339.
- Kobayashi, J.; Cheng, J.; Ishibashi, M.; Wälchli, M. R.; Yamamura, S.; Ohizumi, Y. *J. Chem. Soc., Perkin Trans. 1* **1991**, 1135–1137.